The following errors have been identified in the Organic Chemistry Student Manual, 7th Ed:

p. 3 3.2b) For 3.1d
p. 5 7.1b) $CH_3CH_2CH_2OH$
7.2b) $CH_3CH_2C(=O)OH$
p. 7 3.2 b) answer drawn with wrong angle
p. 19 1.4a) $CH_2=C(CH_3)CH(OH)(CH_2)_2C(CH_3)_2CH_2CO_2H$
p. 22 5. $C_5H_9NO_2$
p. 35 4.3a) $H_2C=CH^-$ (anion)
p. 45 3.1n) *cis*-1-bromohex-3-ene
4.1c) 1-bromo-2-methylcyclobutane
p. 49 13. online answer should be isobutylcyclopentAne
p. 61 5.1e) 4-bromo-3,5-dimethyl-1-hexene
p. 88 5. no answers on answer key. a) nonpolar; b) polar protic; c)polar aprotic; d) polar protic
p. 98 1.2c) ClH_2C (delete the 2nd C on top left group)
p. 107 remove cyclopentane ring answers
p. 175 2.1b) 2 ^1H NMR and 3 ^{13}C signals
p. 186 4.2e) Ph-CH_2-CH_2-CH_2OH from toluene. (Delete last step in answers on p. 191)
p. 191 4.2d) first reagent should be PBr_3; HBr will also add to pi bond.
p. 198 5 d, e, f): add heat
p. 205 2c) add heat
p. 245 7.2e) change second step #2 to excess CH_3OH, H+
p. 250 7.2d) Answer should be E2 products, 2-methyl-2-butanol + $CH_2=CH_2$
p. 251 7.2f) alkyne has extra C
p. 270 1.2a) $(CH_3)_3\mathbf{C}CH_2COOH$
p. 271 3.2a) change Cl to CH_3
p. 273 5c) change Cl on reactant to OH to match answer key
p. 276 1.2b) ethyl propanoate
p. 276 2.2a) d; 2.2b) e
p. 281 6c) remove one CH_2 from product; even if you use an epoxide to extend the chain by 2C, oxidation conditions would oxidize the benzylic C
p. 289 3.1a) on the product, replace far right CH_3 with a H (should be aldehyde)
p. 303 2c) $(CH_3)_2CHCH_2CH_2NH_3^+$